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Chapter 1: Structure and Bonding Page 7

37. What is the approximate C-C-C bond angle in propene,  $\text{CH}_3\text{CH}=\text{CH}_2$ ? A)  $90^\circ$  B)  $109.5^\circ$  C)  $120^\circ$  D)  $180^\circ$  38. What is the approximate H-C-O bond angle in formaldehyde,  $\text{H}_2\text{CO}$ ? A)  $90^\circ$  B)  $109.5^\circ$  C)  $120^\circ$  D)  $180^\circ$  39. Determine the electron geometry around the indicated atom in each species.

## Chapter 1: Structure and Bonding - test bank U

Structure and Bonding 2 ANS: Exhibit 1-2

Consider the structure of urea, shown below, to answer the following question(s). 8. Refer to Exhibit 1-2. Fill in any non-bonding valence electrons that are missing from the line-bond structure. ANS: 9. Refer to Exhibit 1-2. The carbon atom in urea is: a.  $\text{sp}^3$  hybridized b.  $\text{sp}^2$  hybridized c.  $\text{sp}$  hybridized

## Chapter 1—Structure and Bonding

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Banks for exam solving. ... Chapter 1: Structure and Bonding Chapter 1: Structure and Bonding 1. What is the ground-state electronic configuration of a carbon atom? A)  $1s^2, 2s^2, 2p^5$  B)  $1s^2, 2s^2, 2p^2$  C)  $1s^2, 2s^2, 2p^6$  D)  $1s^2, 2s^2, 2p^4$  ...

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(1) In the  $\beta$  sheet structure, several extended polypeptides, or two regions of the same polypeptide, lie side by side and are stabilized by hydrogen bonding

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between adjacent chains. (2) Adjacent chains may be either parallel (with a repeat distance of about 6.5 Å) or antiparallel (7 Å repeat).

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6. A hydronium ion: A) has the structure  $\text{H}_3\text{O}^+$ . B) is a hydrated hydrogen ion. C) is a hydrated proton. D) is the usual form of one of the dissociation products of water in solution.

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Test Bank Organic Chemistry, 7th Edition Paula Yurkanis Bruice Organic Chemistry, 7e (Bruice) Chapter 1 Remembering General Chemistry: Electronic Structure and Bonding . 1) Atoms with the same number of protons but different numbers of neutrons are called \_\_\_\_\_. Answer: isotopes. Section: 1-1

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Answer: These are [...]

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bank help me in school? Think about it like this. You have one text book in your class. So does your teacher. Each text book has one test bank that teachers use to test students with.

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Unit 13 - Electrochemistry

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Structure and Synthesis of Alkenes 1)  
Which of the following statements best describes the relative bond dissociation energies of the sigma and pi bonds present in the carbon-carbon double bond of an

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