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Book Solutions Lehninger Biochemistry Chapter 17-Fatty ...

Lehninger Principles of Biochemistry 6th Edition Nelson Solutions Manual. 5-14. 1. Solubility of Ethanol in WaterExplain why ethanol (CH₃CH₂OH) is more soluble in water than is ethane (CH₃CH₃). AnswerEthanol is polar; ethane is not.

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Chapter 2 Water 5-15 5. Calculation of Hydrogen Ion Concentration from pHWhat is the H⁺ concentration of a solution with pH of (a) 3.82; (b) 6.52; (c) 11.11? Answer Using [H⁺] 10^{-pH}: (a) [H⁺] 10^{-3.82} 1.51 10⁻⁴ M; (b) [H⁺] 10^{-6.52} 3.02 10⁻⁷ M; (c) [H⁺] 10^{-11.11} 7.76 10⁻¹² M. 6. Acidity of Gastric HClIn a hospital laboratory, a 10.0 mL sample of gastric juice, obtained several

Water - 2020 Test Bank and Solutions Manual

See an explanation and solution for Chapter 13, Problem 14 in Cox/Nelson's Lehninger Principles of Biochemistry (7th Edition).

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