

Modern Chemistry Chapter 7 2 Review Answers

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Classification of Organic Compound - 12th Class Chemistry ~~Modern Chemistry Chapter 7 2~~

Modern Chemistry □ CHAPTER 7 HOMEWORK 7-2 (pp. 206–211) VOCABULARY Match the symbol with the name that describes it. 1. Br- _____ a. nitrate 2. OH- _____ b. chlorite 3. NO₂ - _____ c. hydroxide 4. NO₃ - _____ d. bromide 5. O₂ - _____ e. nitrite 6. CN - _____ f. ammonium 7. ClO₂ - _____ g. oxide 8. NH₄⁺ _____ h. cyanide GRAPHIC ORGANIZER

~~HOMEWORK 7-2~~

Ba(OH)₂ —The subscript 2 refers to everything inside the parentheses, so there are two hydroxide ions for every atom of barium. (NH₄)₂ SO₄ HOMEWORK 7-2 Modern Chemistry Chapter 7 Section 2 Review. 7 terms. Modern Chemistry Chapter 7 Section 3. 4 terms. Modern Chemistry Chapter 7 Section 4. 30 terms. Polyatomic Ions. Features. Quizlet Live.

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1) Calculate the amount of each element in sample. 2) Convert the mass composition of each element in moles. %/mass divide smallest number. If given the % of each element in a compound, be able to calculate the empirical formula of the compound. %/ element mass; divide answers by the smallest number.

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$2 = 0.947 \text{ atm}$; $P_2 \text{ of } O_2 = 0.987 \text{ atm}$ Unknown: $V_2 \text{ of } O_2 \text{ in mL}$ VPLAN Rea r ange the equ tion for Boyl 's l w ($P_1 V_1 = P_2 V_2$) to obtain V_2 . $V_2 = P_1 V_1 / P_2$ SSOLVE ubstitute values for P_1 , V_1 , and P_2 to obtain the new volume, V_2 . $V_2 = 2 (0.947 \text{ atm})(150.0 \text{ mL } O_2) / 0.987 \text{ atm} = 144 \text{ mL } O_2$ CHECK YOUR WORK When the p ress ue is inc ea sd ...

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~~Chapter 3 — Chemistry~~

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Title: Modern Chemistry Chapter 7 Chemical Formulas 1 Modern Chemistry Chapter 7 Chemical Formulas Chemical Compounds . A chemical formula indicates the kind and relative number of atoms in a chemical compound. C₈H₁₈ (octane) has 8 carbon and 18 hydrogen atoms. 2 Forming Ionic Compounds. Compounds that have the elements held together by

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