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distinct type of molecular compound;  
would show up on pH scale as a value  
below 7 Binary Acids Are acids that  
consist of two elements, usually  
hydrogen, and one of the halogens.

**7 Chemical Formulas and Chemical**

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continued Class Date 6. How many total moles of atoms are there in one mole of silver(I) phosphate,  $\text{Ag}_3\text{P}_2\text{O}_8$ ? d. 11 7.

What is the correct molar mass of  $\text{Ag}_3\text{P}_2\text{O}_8$ ? ... Assessment Class Date  
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Compounds Section Q Iz: Determining  
Chemical Formula

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Modern Chemistry 85 Chapter Test

Name \_\_\_\_\_ Class \_\_\_\_\_ Date \_\_\_\_\_ Chapter Test A,

continued \_\_\_\_\_ 6. Which factor is the most important in determining the average kinetic energy of gas particles?

- a. pressure b. temperature c. volume of the container d. mass of the container

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**Assessment Chapter Test A -  
Kettering City School District**

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Chapter Test Assessment Chapter Test A  
Chapter: The Periodic Law Use the  
periodic table below to answer the  
questions in this Chapter Test. In the

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space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_\_ 1.

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CHAPTER 7 REVIEW Chemical Formulas  
and Chemical Compounds MIXED

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REVIEW SHORT ANSWER Answer the following questions in the space provided.

1. Write formulas for the following compounds:

CuCO<sub>3</sub> a. copper(II) carbonate  
Na<sub>2</sub>SO<sub>3</sub> b. sodium sulfite  
(NH<sub>4</sub>)<sub>3</sub>PO<sub>4</sub> c. ammonium phosphate  
SnS<sub>2</sub> d. tin(IV) sulfide  
HNO<sub>2</sub> e. nitrous acid

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Modern Chemistry Chapter 7 test.

STUDY. PLAY. The basis of an ionic bond is the\_\_ absorption of water into their solid structures. Elements tend to gain or lose electrons in order to acquire the electron configuration of a\_\_ noble gas. Ions that are formed from more than one atom are called\_\_



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54 CHEMICAL FORMULAS AND CHEMICAL  
COMPOUNDS MODERN CHEMISTRY ...  
CHAPTER 7 REVIEW Chemical Formulas  
and Chemical Compounds SECTION 2  
SHORT ANSWER Answer the following  
questions in the space provided. 1. ... 6.  
a.? 7 Chemical Formulas and Chemical  
Compounds 1. 2 b. 4) 3) b.) ...

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binary compounds. nomenclature.

## **7 Chemical Formulas and Chemical Compounds**

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... CH 7 Pre Test . White Board Practice.  
CH 7 Ionic & Covalent Nomenclature.

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Chapter Test Assessment Chapter Test B  
Chapter: The Periodic Law PART I In the  
space provided, write the letter of the

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term or phrase that best completes each statement or best answers each question. \_\_\_\_ 1. In his periodic table, Mendeleev did not list all of the elements in order

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Modern Chemistry 68 Chapter Test

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Name \_\_\_\_\_ Class \_\_\_\_\_ Date \_\_\_\_\_ Chapter Test A,  
continued \_\_\_\_\_ 21. Metal X replaces the  
ions of metal Y from solution, but it  
cannot replace the ions of metal Z from  
solution. The order these metals should  
have in the activity series (from top to  
bottom) is a. x, y, z b. z, x, y

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Modern Chemistry 91 Chapter Test  
Name Class Date Chapter Test B,  
continued 27. Distinguish between ionic  
crystals and metallic crystals. PART V

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Write the answers to the following questions on the line to the left, and show your work in the space provided. The molar enthalpy of fusion for water is 6.008 kJ/mol. 28.

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