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(Question 2 continued) (d)A chemist considered preparing a copper(I) salt by reacting copper metal with the corresponding copper(II) salt according to the equation below.  $\text{Cu}(\text{aq}) + \text{Cu}(\text{s}) \rightarrow 2\text{Cu}(\text{aq})^{2+}$   
(i)Using data from Table 14 of the Data Booklet, calculate the cell potential for this reaction.

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electronic transition would absorb the  
radiation of the shortest wavelength?  $n = 5$   $n = 4$   $n = 3$   $n = 2$   $n = 1$  AB CD 5.  
Which is ...

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