

## Chemistry If8766 Structure Of Hydrocarbons

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### Chemistry If8766 Structure Of Hydrocarbons

STRUCTURE OF HYDROCARBONS Draw the structure of the compounds below. J Chemistry IF8766 97 ©Instructional Fair, Inc, 1. ethane 5.  $t\text{-L-C}_5\text{-4c}^*6\text{Lm-r-+h}$ , (jl h,eP vie, -C C C C C C.- 0- 2. ~ al W'~-' I il~~ s'-C- -C s - ff 6. 3, 3-dimethyl pentane  $Y\sim I\_v( 3 \text{ Li } , -\text{cl} i e\sim\text{In}\sim\text{I} - 3, 3 -\text{d} i \text{rn}\sim 4\text{ll}\sim\{ \text{Oc.}+\text{cLn} e v 1. I \sim \text{G} \text{IC} - \text{I} i 1$

### NAMING HYDROCARBONS Name - Ms Galloway

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Hydrocarbons with the same formula, including alkanes, can have different structures. For example, two alkanes have the formula  $C_4H_{10}$ : They are called n-butane and 2-methylpropane (or isobutane), and have the following Lewis structures:. The compounds n-butane and 2-methylpropane are structural isomers (the term constitutional isomers is also commonly used).

## 20.1 Hydrocarbons - Chemistry

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methane enstructi( 97 Chemistry IF8766. NAMING HYDROCARBONS me the compounds below according to the IUPAC naming system c— c- c- c- c- c- c- c- Name H H H H C —C —C—H H c- H H H CH3 H c— d H H H H H H H H H H Chemistry IF8766 H H H H H c- H H H H c- CH3 c- H H 96  
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## Chemistry If8766 Answers Page 91

Unsaturated hydrocarbons have one or more double or triple bonds between carbon atoms. Those with double bond are called alkenes and those with one double bond have the formula  $(C_nH_{2n})$  (assuming non-cyclic structures). Those containing triple bonds are called alkynes, with general formula  $(C_nH_{2n-2})$ . The smallest alkene—ethene—has two C atoms and is also known by its common name ...

## 25.2: Introduction to Hydrocarbons - Chemistry LibreTexts

- Saturated hydrocarbons - contain only C & H and no multiple bonds
- Non-cyclic alkanes have the formula  $C_nH_{2n+2}$
- 8 H C H H H methane  $CH_4$  propane  $C_3H_8$  pentane  $C_5H_{12}$  cyclopentane  $C_5H_{10}$
- Alkanes are a little dull - used as solvents or burnt as fuel
- Methane is a fuel, a chemical feedstock & a green house gas propane  $CH_4$

## Unit One Part 2: naming and functional groups

Hydrocarbon, any of a class of organic chemical compounds composed only of the elements carbon

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(C) and hydrogen (H). The carbon atoms join together to form the framework of the compound, and the hydrogen atoms attach to them in many different configurations. Hydrocarbons are the principal constituents of petroleum and natural gas.

### **hydrocarbon | Definition, Types, & Facts | Britannica**

The purpose of organic chemistry nomenclature is to indicate how many carbon atoms are in a chain, how the atoms are bonded together, and the identity and location of any functional groups in the molecule. Name m, e, p, b STRUCTURE OF HYDROCARBONS Draw the structure of the compounds below.

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This molecule has a unique structure, as its methoxy groups are located internally rather than at its periphery. "Initially, we wanted to see if this hydrocarbon demonstrated novel phenomena due ...

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In the lab, they synthesized an aromatic hydrocarbon called methoxy-substituted as-indacenoterrylene. This molecule has a unique structure, as its methoxy groups are located internally rather than ...

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