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Updated 3/30/16 Lots

and lots of significant

figures! Significant

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worksheet: You'll be

significantly figuring in

no time! Significant

figures calculations

sheet: Do calculations

using the magic of

significant figures!

Significant Figures I (dd-

ch): Practice finding

how many significant

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Significant Figures Practice

Practice worksheets | The Cavalcade o' Chemistry

Rule 2: When you multiply or divide numbers, the answer should be written with the same number of significant figures as the number with the fewest significant figures. In our example, our answer should be written as "0.1 g/mL" because

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both 10 grams and 90 mL have one significant figure.

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Significant Figures I (dd-ch): Practice finding how many significant figures a measured value has. Also helps you to figure out what significance you get from sig figs in the first place. Significant Figures II (dd-ch):

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Similar to the sheet above, but with different problems and a slightly different format. Significant Figures V (dd-ch): Practice doing significant figure calculations. This should really help with your sig fig knowledge.

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Significant Figures

Practice Answers

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Chemfiesta **Significant Figures** **Practice Answers**

Practice Problems:

Significant Figures

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(Answer Key) How many Significant figures in each term? a. 34.6209 6 b. 0.003048 4 c. 5010.0 5 d. 4032.090 7; Solve the following equations using the correct number of significant figures. a. $34.683 + 58.930 + 68.35112$ 161.964 b. $45001 - 56.355 - 78.44$ 44866

Practice Problems: Significant Figures (Answer Key)

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answers hence simple!
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Answers

3. $8.888 \text{ m} \times 3.29853$

$\text{m} = 29.32 \text{ m}^2$ The

answer must have 4

sig dig like 8.888. 4.

$1.25 \text{ mm} + 3.2 \text{ mm} =$

4.5 mm The answer

can only have one

decimal place like the

3.2. 5. $145 \text{ L} - 0.222 \text{ L}$

$= 145 \text{ L}$ The answer

cannot have any

decimal places cause

145 has none. 6. 145 g

$- 0.99 \text{ g} = 144 \text{ g}$ The

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Practice Problems for
Significant Figures .

RULE -1: If the decimal
is . Present: Find the
first non zero on the
left, then count all
digits to the . RIGHT. If
the decimal is . Absent:
Find the first non zero
on the right, then
count all digits to the .
LEFT. RULE-2: Every
digit in scientific
notation is Significant

Significant Figures

Practice Worksheet

Addition and subtraction with significant figures Our mission is to provide a free, world-class education to anyone, anywhere. Khan Academy is a 501(c)(3) nonprofit organization.

Significant figures (practice) | Khan Academy

Q. Calculate $345.009 \text{ g} - 23.00 \text{ g}$ and give your answer with the correct

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number of significant figures. Q. Calculate $12.34 + 1.234 + 0.1234$ and give your answer with the correct number of significant figures. Q. Calculate $5.50 \text{ cm} + 5.5 \text{ cm}$ and give your answer with the correct number of significant figures.

Significant Figures Practice | Other Quiz - Quizizz

Practice worksheets. A.
Scientific method &

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graphing (3) B. Unit conversions (4) C. Significant figures (6) D. Atomic theory & calculations (6) E. The periodic table (4) F. Ionic compounds (6) G. Naming compounds (18) H. Lewis structures & VSEPR (4) I. IM forces, polarity, & metals (7) J. Moles & molar masses (7) K. Percent composition (2) L ...

Worksheets! | The
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**Cavalcade o'
Chemistry**

Sometimes it seems as if the world is falling apart. COVID-19 has forced us out of school and into hiding. The murder of George Floyd and subsequent brutality has left us confused and afraid because the society in which we live has let us down in important ways. There are a lot of things in...

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Celebrating 20 years

of ...

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Practice Problems

Answers The short

answer: Stoichiometry

is how you figure out

how much stuff will be

made in a chemical

reaction, or how much

stuff you'll need to use

when performing a

chemical reaction The

calculations that make

this

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Report the answers to
the correct number of
significant figures: 51)

$$(10.3 + 0.01345) \div$$
$$(10.3 \times 0.01345) =$$

$$74.4 \quad 52) (155.234 \times$$
$$0.789) + (154.2 \times$$

$$0.120) + (132.8 \times$$
$$0.0901) = 153 \quad 53) (234$$

$$\times 0.000123) + (234 \times$$
$$0.999) = 234$$

Practice Worksheet for Significant Figures

The measurement,
20600 molecules, has

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Significant
Figures Practice
Answers

how many significant digits? 1. 2. 3. 4. 5. Add the following three numbers and report your answer using significant figures: $2.5 \text{ cm} + 0.50 \text{ cm} + 0.055 \text{ cm} = ?$ 3.055 cm.

Mathematics Using Significant Figures Quiz

Answer. The calculator answer is 921.996, but because 13.77 has its farthest-right significant figure in the

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hundredths place we need to round the final answer to the hundredths position. Because the first digit to be dropped (in the thousandths place) is greater than 5, we round up to 922.00.
 $922.00 = 9.2200 \times 10^2$.

2.4: Significant Figures in Calculations - Chemistry ...

Significant figures have

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Significant
Figures Practice
Answers

been taken into
account in all answers.

1) Convert 3.4 liters of
milliliters. 3400 mL, or
 $3.4 \times 10^3 \text{ mL}$. 2)

Convert 876
millimeters to meters.

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cd98f00b204e9800998
ecf8427e.