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12CHAPTER Assessment Reviewing Content Assessment 379 12.1 The Arithmetic of Equations 36. Interpret each chemical equation in terms of interacting particles. a. $2\text{KClO}_3(s) \rightarrow 2\text{KCl}(s) + 3\text{O}_2(g)$ b. $4\text{NH}_3(g) + 6\text{NO}(g) \rightarrow 2\text{N}_2(g) + 6\text{H}_2\text{O}(g)$ c. $4\text{K}(s) + \text{O}_2(g) \rightarrow 2\text{K}_2\text{O}(s)$ 37. Interpret each equation in Problem 36 in terms of interacting numbers of moles of reactants and ...

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