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Solids - 12.2 Crystalline & Amorphous Solids - crystalline vs. amorphous solids -- crystalline solid: cubic array of tightly packed atoms or molecules --- ionic solids - NaCl, comprised of an anion and a cation

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Chapter 12 - Solids and Modern Materials

Chapter 12 Solids and Modern Materials Professor Olga Chumpitaz 1) A solid has a very high melting point, great hardness, and poor electrical conduction. This is a(n) solid. A) ionic B) molecular C) metallic D) covalent network E) metallic and covalent network 2. All of the following are a type of solid except

Solved: Chapter 12 Solids And Modern Materials Professor O ...

Ch. 12: Crystalline Solids and Modern Materials Dr. Namphol Sinkaset Chem 200: General Chemistry I . I. Chapter Outline I. Introduction II. X-Ray Crystallography III. Unit Cells and Basic Structures IV. Types of Crystalline Solids V. Semiconductors and Band Theory . I. Materials Science

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Chapter 12: Solids and Modern Materials includes 118 full step-by-step solutions. This textbook survival guide was created for the textbook: Chemistry: The Central Science, edition: 12. Chemistry: The Central Science was written by and is associated to the ISBN: 9780321696724. Since 118 problems in chapter 12: Solids and Modern Materials have ...

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12.7: Solids and Modern Materials (Exercises) These are homework exercises to accompany the Textmap created for "Chemistry: The Central Science" by Brown et al. Thumbnail: The alpha form of solid polonium.

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Chapter 12 - Structures of Solids: Part 2 of 6

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CHAPTER 12: Solids and Modern Materials CHEM:1110

(Strathman) Fall 2018 2 Unit Cells and Crystal Lattices 2-D

Examples • think of “tiles” covering entire space • 2 lengths, 1

angle • dots = lattice points and corners of cell • Only five 2-D

options: square, rectangle, rhombus, oblique, hexagon 4 3-D

Examples of real crystals • 7 types of 3-D unit cells • 3 lengths &

3 angles ...

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